

SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR  
(AUTONOMOUS)

B.Tech I Year II Semester Regular & Supplementary Examinations May/June-2026  
CHEMISTRY

(Common to EEE, ECE, CSE & CSIT)

Time: 3 Hours

Max. Marks: 70

**PART-A**

(Answer all the Questions 10 x 2 = 20 Marks)

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|---|---|---|-----|----|----|
| 1 | a | State the significance of $\psi$ and $\psi^2$ in quantum mechanics. | CO1 | L1 | 2M |
|   | b | Define bond order. Calculate bond order of $O_2$ molecule.          | CO1 | L2 | 2M |
|   | c | Define semiconductors and give one example.                         | CO2 | L1 | 2M |
|   | d | List two properties of carbon nanotubes.                            | CO2 | L2 | 2M |
|   | e | State Nernst equation for a cell reaction.                          | CO3 | L1 | 2M |
|   | f | Mention two advantages of hydrogen-oxygen fuel cells.               | CO3 | L2 | 2M |
|   | g | Define functionality of a monomer with an example.                  | CO4 | L1 | 2M |
|   | h | What are Biodegradable polymers?                                    | CO4 | L2 | 2M |
|   | i | State Beer-Lambert's law.   | CO5 | L1 | 2M |
|   | j | What is the principle of HPLC?                                      | CO5 | L2 | 2M |

**PART-B**

(Answer all Five Units 5 x 10 = 50 Marks)

**UNIT-I**

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|---|--|--|-----|----|-----|
| 2 |  | Illustrate the molecular orbital diagram of $O_2^+$ and $O_2^{2-}$ . Explain its bond order and magnetic property based on MOT theory. | CO1 | L2 | 10M |
|---|--|--|-----|----|-----|

OR

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|---|---|--|-----|----|----|
| 3 | a | Explain Planck's Quantum Theory.                     | CO1 | L2 | 5M |
|   | b | Write short notes on Wave-Particle duality of matter | CO1 | L2 | 5M |

**UNIT-II**

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|---|---|--|-----|----|----|
| 4 | a | Explain the basic principle and Classifications of Super Capacitors. | CO2 | L2 | 5M |
|   | b | Discuss applications of Super Capacitors.                            | CO2 | L2 | 5M |

OR

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|---|---|---|-----|----|----|
| 5 | a | What is meant by Nano materials? How the Nano materials Classified. | CO2 | L1 | 5M |
|   | b | Write a note on applications of fullerenes.                         | CO2 | L1 | 5M |

**UNIT-III**

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|---|---|--|-----|----|----|
| 6 | a | Define Electrochemical cell? Explain the construction, working Principle and mechanism of an Electrochemical cell. | CO3 | L1 | 5M |
|---|---|--|-----|----|----|

- |  |   |   |     |    |   |
|--|---|---|-----|----|---|
|  | b | Calculate the emf of iron-copper voltaic cell $[Fe/Fe^{2+}/Cu^{2+}/Cu]$ with standard potential of copper and iron as + 0.34 V and - 0.44 V respectively. | CO3 | L3 | 5 |
|--|---|---|-----|----|---|

OR

- |   |   |   |     |    |   |
|---|---|---|-----|----|---|
| 7 | a | What is primary Battery? Write about construction, cell reactions and applications of Zinc-Air battery. | CO3 | L1 | 6 |
|   | b | Differentiate Primary and Secondary Batteries with examples.  | CO3 | L2 | 4 |

**UNIT-IV**

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|---|---|---|-----|----|---|
| 8 | a | Write the preparation, properties and application of Buna-S rubber and Buna-N rubber. | CO5 | L2 | 6 |
|   | b | Write the applications of conducting polymers.  | CO5 | L3 | 4 |

OR

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|---|---|--|-----|----|---|
| 9 | a | Discuss the synthesis, properties and applications of Polyvinylchloride (PVC) polymer. | CO5 | L2 | 5 |
|   | b | Distinguish between Thermoplastics and Thermosetting plastics.                         | CO5 | L2 | 5 |

**UNIT-V**

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|----|---|---|-----|----|---|
| 10 | a | Explain the different regions of electromagnetic spectrum.    | CO6 | L2 | 5 |
|    | b | Discuss the basic components of UV-Visible spectroscopy.      | CO6 | L1 | 5 |
| 11 | a | Explain various classifications of Chromatographic technique. | CO6 | L2 | 5 |
|    | b | Explain in detail about Stretching and bending vibrations.    | CO6 | L2 | 5 |

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